KAA Curriculum Ove	rview	Science	Year	10	EOY Exam	Sequencing and Progression	
Rationale			I		What content and skills will be	How does this year build on what	How will it benefit them as they
Give an overview of what students are studying this year and why. Link directly to your overall curriculum intent. Students start their GCSE science course this year. In physics, students will study energy and electricity. In biology, students will infection and response, bioenergetics and homeostasis. In chemistry, students will study quantitative chemistry, reactions and organic chemistry. The curriculum will give the students both substantive and disciplinary knowledge that they need to understand and explain phenomena that they experience in their everyday lives. The students will know more and be able to explain more over time. They will also be encouraged to think for themselves and to be curious and analytical as they look at experimental data.					assessed in the EOY exam? Students will be assessed in the form of 3 full paper 1 GCSE combined science exams	<i>they've learnt last year?</i> -Directly links to each biology, chemistry and physics topic taught at KS3	move forward next year? This curriculum builds on the year 9 core principles and focusses on topics in more detail. In year 11, students will apply knowledge from year 10 with increasing demand from the content.
Term	Autumn 1		Autumn 2	Spring 1	Spring 2	Sum 1	Sum 2
Link to MTP Overview	Aut1 MTP						
Topic studied & Fertile Question		ogy: Infection and & Bioenergetics	GCSE Chemistry: Quantitative Chemistry, Chemical Reaction Energy Changes		GCSE Biology: Homeostasis and response	GCSE Chemistry: Organic Chemistry, Analysis & the Earth's Resources	GCSE Physics: Electricity
Adjustments following last assessments / evaluation.	None		None	None	None	None	None
Key knowledge and skills students need to have gained by the end of the unit	 Defining a identifying bacteria, v Diseases bacteria, fu Structure and anima Transmiss non-spec systems: fi Specific h antibody p antitoxin p vaccinatio Antibiotio Discovery drugs Clinical tr Bioenerget Ph Ph en Fau 	ind response a pathogen and the four pathogens: irus, protist, fungi caused by viruses, ungi and protist. of eukaryotic (plant l) and prokaryotic cells sion of pathogens ific human defence rst lines of defence numan defence system: production and production on and herd immunity cs versus painkillers y and development of rials and ethics	 Knowledge Quantitative chemistry Defining the conservation of mass and using the concept to balance chemical equations. Defining relative formula ma (RFM) and using the periodic table to calculate the RFM of compounds. The production of gas can reduce the mass of products i an open system, as it is lost to the surroundings Estimate uncertainty Calculate moles using mass/ 1 mole of a substance in gra is equal to its RFM 1 mole of substance contain 6.02 x 10²³ particles (Avogadr constant). Calculating reacting masses using balanced symbol equati in excess. Using balanced symbol equations to identify the limit reactant. Calculating concentrations in g/dm³. 	 Energy stores include: GPE, kinetic energy, electrostatic, thermal, elastic, nuclear, and chemical. Energy transfers: heating, forces, current. Difference between energy stores and transfers. kinetic energy = 0.5 × mass × (speed)² elastic potential energy = 0.5 × spring constant × (extension)² g. p. e. = mass × gravitational field strength × height change in thermal energy = 0.5 × mass × specific heat capacity × temperature change power = energy transferred/time Energy is not created or destroyed it is transferred between the system and surroundings. Efficiency = useful output 	 Knowledge Define homeostasis Describe how homeostasis maintains the bodies internal conditions The structure of the nervous system and how it is adapted to its function. The reflex arc and its automatic response Human endocrine system and its role Human endocrine system as a network of glands which secrete hormones directly into the bloodstream to reach target organs. Identify the position of the: pituitary gland, pancreas, thyroid, adrenal gland, ovary and testes. Control of blood glucose concentration by the pancreas Type 1 and 2 diabetes and its treatment Hormones in reproduction The menstrual cycle Contraception 	Knowledge Organic - Crude oil as a finite resource formed from the remains of ancient biomass - Alkanes and their properties - Alkanes and their properties - Fractional distillation - Cracking – steam and catalytic and the conditions required - Properties of hydrocarbons: boiling point, viscosity, flammability Chemical analysis	 Knowledge Use circuit symbols to draw series and parallel circuit diagrams Describe the difference between series and parallel circuits and how the current and potential difference varies. Current is the rate of flow of charge measured by an ammeter. Charge flow = current x time Potential difference = current x resistance Draw graphs to show resistance from LDR, lamp and a resistor at constant temperature Resistance in series versus parallel Direct and alternating current Mains electricity and the 3 pin plug.

	- Uses of glucose from		- Describe the main energy	- The use of hormones to treat	Chemistry of the atmosphere	- Calculating power =
	photosynthesis	Chemical changes	sources available on earth and	infertility	- Describe how the Earth's	potential difference x
	- Aerobic vs anaerobic	- Using various	the environmental impacts	- Feedback systems	early atmosphere formed	current. Power=
	respiration	experiments to identify	- Renewable energy alternatives		- The composition of the	(current) ² x resistance
	- Products of anaerobic	the reactivity series	to current fossil fuels used.		Earth's current	- Energy transferred =
	respiration in humans,	(metal + acid, metal +		Skills	atmosphere	power x time. Energy
	plants and yeast cells	water, metal carbonate +		 Extract and interpret 	- Describe and explain	transferred = charge flow
	- Response to exercise.	acid, metal oxide+ water)		data from graphs, charts	how oxygen increased	x potential difference
	How the human body	- Define oxidation and	Skills	and tables	and carbon dioxide	- The national grid
	reacts to the increased	reduction in terms of	- RP14: an investigation to	- RP6 : Plan and carry out	decreased	
	demand for energy	gain of oxygen/loss of	determine the specific	an investigation into the	- Greenhouse gases and	
	- Metabolism	oxygen and loss of	heat capacity of one or	effect of a factor on	the greenhouse effect	Skills
	Wittabolishi	electrons/gain of	more materials. The	human reaction time	- How human activities	- RP15: use circuit
	Skills	electrons	investigation will involve	- Evaluate information	contribute to the	diagrams to set up and
	Infection and response	- Use the reactivity series	linking the decrease of	around the relationship	greenhouse effect	check appropriate
	- Evaluation of data	to identify methods of	one energy store (or	between obesity and	- Carbon footprint and	circuits to investigate the
	presented in graphs and	extraction of metals from	work done) to the	diabetes, and make	lifecycle assessments	factors affecting the
	tables looking at global	their ores.	increase in temperature	recommendations taking	- Atmospheric pollutants	resistance of electrical
	use of vaccinations to	- Write ionic equations for	and subsequent increase	into account social and		circuits. This should
	prevent disease.	displacement reactions,	in thermal energy stored.	ethical issues.	Using resources	include:
		identifying what has	- Confidently rearrange	 Interpret and explain 	- Potable water from fresh	• the length of a wire at constant
	- Interpreting clinical trials	been oxidised and what	equations	simple diagrams of	water and waste water	temperature
	graphs and data	has been reduced.	- Evaluate use of different	negative feedback	- Bioleaching and	Combinations of resistors in
		- Neutralisation reactions	energy sources.	control	phytoextraction from low	series and parallel.
		to produce soluble salts		Evaluate from the	grade ore	- RP16: use circuit
	Bioenergetics	and water		perspective of patients	- Reduce, reuse, recycle	diagrams to construct
	- Measure and calculate	- pH scale and		and doctors the methods		appropriate circuits to
	the rates of	neutralisation. Recall		of treating infertility	Skills	investigate the I–V
	photosynthesis	difference between a			- Opportunity to develop	characteristics of a
	- Extract and interpret	weak and strong acid.			practical skills through:	variety of circuit
	graphs of photosynthesis	- Electrolysis in molten			Investigating the properties of	elements, including a
	rate involving 1 limiting	and aqueous solutions			different hydrocarbons.	filament lamp, a diode
	factor	- Using half-equation to			- Use data to evaluate the use of	and a resistor at constant
	- RP5: Investigate the	represent the reactions			various fractions produced from	temperature.
	effect of light intensity	at the electrodes.			factional distillation.	 Rearrange and apply
	on the rate of	- Electrolysis of aluminium			- RP12: Investigate how paper	equations.
	photosynthesis using an	from aluminium oxide			chromatography can be used to separate and tell the difference	
	aquatic organism such as	dissolved in molten			between coloured substances.	
	pondweed.	Cryolite.			Students should calculate Rf	
		,			values.	
					-RP13: Analysis and purification	
					of water samples from different	
		Skills:			sources, including pH, dissolved	
		- Using ratios to balance			solids and distillation.	
		equations and calculate			- Evaluate products through	
		reacting masses			lifecycle assessments	
		- RP8: Preparation of a			- Translate information between	
		pure, dry sample of a			graphical and numeric form.	
		soluble salt from an				
		insoluble oxide or				
		carbonate, using a				
		Bunsen burner to heat				
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		 dilute acid and a water bath or electric heater to evaporate the solution. RP9: Investigate what happens when aqueous solutions are electrolysed using inert electrodes. This should be an investigation involving developing a hypothesis. 				
How is understanding assessed at the end of the unit?	End of topic test	End of topic test	End of topic test	End of topic test	End of topic test	End of topic test