

## Year 11 Science Half-term Revision

Over half-term, your focus is to **embed the core knowledge** you will be assessed on during your mocks in AUT2. This will cover a **full paper 1 for biology, chemistry and physics**.

There will be an **additional topic of 'forces' in the physics paper**.

### Mocks topics

<b>Biology</b>	<b>Chemistry</b>	<b>Physics</b>
<b>Paper 1:</b> <ul style="list-style-type: none"> <li>• Cell biology</li> <li>• Organisation</li> <li>• Infection and response</li> <li>• Photosynthesis</li> <li>• Respiration</li> </ul>	<b>Paper 1:</b> <ul style="list-style-type: none"> <li>• Atomic structure and the periodic table</li> <li>• Bonding, structure and the properties of matter</li> <li>• Quantitative chemistry</li> <li>• Chemical changes</li> <li>• Energy changes</li> </ul>	<b>Paper 1:</b> <ul style="list-style-type: none"> <li>• Energy</li> <li>• Electricity</li> <li>• Particle model of matter</li> <li>• Atomic structure</li> </ul> <b>Forces</b>

### Tasks

- 1) You must complete **800 points** of Tassomai over the two weeks
- 2) Create revision cards answering the following core questions, using **Tassomai** or **BBC bitesize**

### Key resources

<https://www.bbc.co.uk/bitesize/examspecs/z8r997h>

<https://www.youtube.com/@Freesciencelessons>

<b>Biology</b>	<b>Chemistry</b>	<b>Physics</b>
<ol style="list-style-type: none"> <li>1) State 3 extra organelles a plant cell has compared to an animal cell.</li> <li>2) Order from smallest to largest: organisms, tissue, cell, organ, organ system</li> <li>3) Describe diffusion, active transport and osmosis</li> <li>4) Name 1 example of a specialised cell and its adaptations</li> <li>5) Define pathogen. State the 4 types of pathogens</li> <li>6) Describe what a vaccine contains.</li> <li>7) Equation for photosynthesis</li> <li>8) What factors affect the rate of respiration?</li> <li>9) Equation for respiration</li> <li>10) What is the difference between aerobic and anaerobic respiration?</li> </ol>	<ol style="list-style-type: none"> <li>1) What are the subatomic particles in an atom. State their charge and mass</li> <li>2) Describe the plum pudding model and gold foil experiment</li> <li>3) Draw the electronic structure of Na and state what the group and period number tell you.</li> <li>4) Describe and explain the reactivity of the alkali metals as you go down the group.</li> <li>5) State what ionic bonding occurs between</li> <li>6) Explain why NaCl has a high melting point</li> <li>7) Calculate the relative formula mass of CO<sub>2</sub></li> <li>8) Define exothermic and endothermic.</li> </ol>	<ol style="list-style-type: none"> <li>1) State 4 energy stores</li> <li>2) State 3 energy transfers</li> <li>3) State the equation to calculate efficiency</li> <li>4) Describe the movement of particles in a solid, liquid and gas</li> <li>5) State the 3 types of radiation</li> <li>6) Draw the symbol for a cell and battery</li> <li>7) Define potential difference</li> <li>8) Describe how increasing the number of components affects current</li> <li>9) Describe an experiment used to calculate spring constant</li> <li>10) Describe the difference between mass and weight</li> </ol>